

MeOH gradient and 500 ml fractions were collected. The toxic fractions (25 and 26, 712 mg total) eluted with  $\text{CHCl}_3$ - $\text{Me}_2\text{CO}$  (4:1) were combined and further separated by gel filtration on a  $110 \times 2$  cm column of Sephadex LH-20 with  $\text{CHCl}_3$ -MeOH (1:1). The solid (52 mg) eluted after the toxic oil was dissolved in  $\text{CH}_2\text{Cl}_2$ -hexane and 13 mg of (+)- $\alpha$ (S)-butyramido- $\gamma$ -butyrolactone crystallized from the cooled soln as fine white needles, mp  $120$ – $121^\circ$ ;  $[\alpha]_D^{25} = +18.9^\circ$  ( $\text{CHCl}_3$ ,  $c$  0.74); PMR (1:1  $\text{CDCl}_3/\text{C}_6\text{D}_6$ )  $\delta$  0.80 (t,  $J = 7$  Hz, Me), 1.52 (m,  $J = 7$  Hz,  $\text{CH}_2=\text{Me}$ ), 1.92 (t,  $J = 7$  Hz,  $\text{CH}_2=\text{C}=\text{O}$ ), 1.92 (m, C-3 H), 2.13 (dddd,  $J = 2, 6, 9$  and  $12$  Hz, C-3 H), 3.56 (ddd,  $J = 6, 9$  and  $12$  Hz, C-4 H), 3.82 (ddd,  $J = 2, 9$  and  $9$  Hz, C-4 H), 4.26 (ddd,  $J = 6, 9$  and  $12$  Hz, C-2 H), 6.1 (bd,  $J = 6$  Hz, NH);  $\text{C}^{13}$  NMR ( $\text{CDCl}_3$ )  $\delta$  13.6 (Me), 18.8 ( $\text{CH}_2$ ), 30.3 ( $\text{CH}_2$ ), 37.9 ( $\text{CH}_2$ ), 49.0 ( $\text{CH}_2$ ), 66.0 (CH), 173.4 (C=O), 175.4 (C=O); IR ( $\text{CHCl}_3$ )  $\nu_{\text{max}}$  3425, 2920, 2870, 1780,  $1676\text{ cm}^{-1}$ ; MS  $m/e$  (rel. intensity) 171 (11), 153 (7), 143 (73), 71 (88), 43 (100), 28 (95); high resolution mass measurement 171.088690 (calcd for  $\text{C}_8\text{H}_{13}\text{NO}_3$  171.089547).

**Hydrolysis** The natural (+)- $\alpha$ (S)-butyramido- $\gamma$ -butyrolactone (10 mg) in 2N HCl was refluxed for several hr. The odor of *n*-butyric acid was detected. Evapn. of the solvent gave  $\alpha$ -amino- $\gamma$ -butyrolactone HCl which exhibited a PMR spectrum in

$\text{D}_2\text{O}$  that was identical to that of a commercial sample of  $\alpha$ -amino- $\gamma$ -butyrolactone hydrobromide. No optical rotation, however, could be detected.

**Synthesis.** Racemic  $\alpha$ -butyramido- $\gamma$ -butyrolactone prepared from  $\alpha$ -amino- $\gamma$ -butyrolactone hydrobromide following the procedure of ref. [3] melted at  $83^\circ$ . (+)- $\alpha$ (S)-Butyramido- $\gamma$ -butyrolactone prepared from (–)- $\alpha$ (S)-amino- $\gamma$ -butyrolactone hydrobromide [4] had mp  $120$ – $121^\circ$  and  $[\alpha]_D^{25} +21.1^\circ$  ( $\text{CHCl}_3$ ,  $c$  0.18).

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## SPHINGOSINE DERIVATIVES FROM RED ALGAE OF THE CERAMIALES

JOHN H. CARDELLINA and RICHARD E. MOORE

Department of Chemistry, University of Hawaii, Honolulu, HI 96822, U.S.A

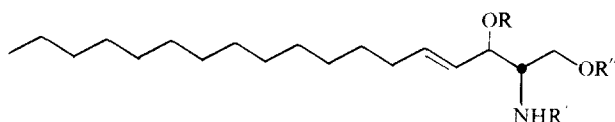
(Revised received 29 July 1977)

**Key Word Index**—*Laurencia nidifica*; *Amansia glomerata*; Rhodomelaceae; dihydrosphingosine *N,O*-diacetate; *N*-acylsphingosines.

**Abstract**—(+)-2(*S*)-*N*-Acetamido-3(*R*)-acetoxyoctadecan-1-ol, a diacetate of dihydrosphingosine, and fatty acid amides of (–)-2(*S*)-amino-3(*R*)-hydroxyoctadec-4(*E*)-en-1-ol (sphingosine) have been isolated from extracts of Hawaiian *Laurencia nidifica* and *Amansia glomerata*, respectively. Although well known as constituents of nerve tissue hydrolyzates throughout the animal kingdom, these compounds have not been previously found in plants.

## INTRODUCTION

Sphingosine (**1a**) has been known since 1882 [1] to be a component of brain tissue hydrolyzate but its structure was not established until 65 years later [2]. This unsaturated amino dialcohol has been found in the hydrolyzates of rat, beef and human brain tissues. In 1941, Lesuk and Anderson [3] isolated dihydrosphingosine (**2a**) from hydrolyzed larvae (*Cysticercus fasciolaris*) of *Taenia taeniaformis*, the common tapeworm of cats. Dihydrosphingosine compounds were later [4] shown to be present in beef spinal cords.

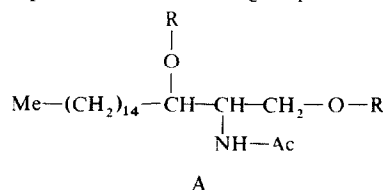


**1a**  $\text{R} = \text{R}' = \text{R}'' = \text{H}$

**1b**  $\text{R} = \text{R}'' = \text{H}$ ,  $\text{R}' = \text{COCH}_2(\text{CH}_2)_n\text{Me}$   
where  $n = 12, 14, 16, 20, 21, 22$

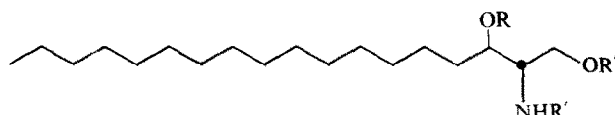
## RESULTS AND DISCUSSION

While examining the more polar extracts of *Laurencia nidifica*, we isolated (+)-2(*S*)-*N*-acetamido-3(*R*)-acetoxyoctadecan-1-ol (**2b**), a diacetate of dihydrosphingosine. The IR spectrum provided evidence for alcohol, ester, and amide functionalities and the PMR spectrum indicated that these 3 groups were at one end of an aliphatic chain. Spin-spin decoupling experiments suggested partial structure **A**. The PMR spectrum of the triacetate derivative indicated a paramagnetic shift of a two proton multiplet (3.58 in **2b**  $\rightarrow$  ca 4.2 in **2c**), fixing the position of the OH group as shown in **2b**.



The MS of **2b** showed fragment ions at  $m/e$  354 for loss of  $\text{CH}_2\text{OH}$  from the  $\text{M}^+$  and  $m/e$  102 for  $\text{AcNH}=\text{CH}-\text{CH}_2\text{OH}$ . The physical constants of the corresponding triacetate (**2c**) were comparable with previously published values [4].

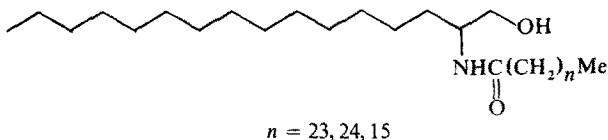
A mixture of related compounds was found in the polar extracts of *Amansia glomerata*. PMR and MS data indicated that these compounds were amides. Hydrolysis led to sphingosine and a mixture of unbranched, saturated fatty acids. The sphingosine was converted to the triacetate and identified by direct comparison with an authentic sample. MS analysis



- 2a**  $\text{R} = \text{R}' = \text{R}'' = \text{H}$   
**2b**  $\text{R}'' = \text{H}, \text{R} = \text{R}' = \text{Ac}$   
**2c**  $\text{R} = \text{R}' = \text{R}'' = \text{Ac}$

of the fatty acid showed that it was composed of predominantly  $n\text{-C}_{23}$ ,  $\text{C}_{24}$  and  $\text{C}_{25}$  acids with smaller amounts of  $\text{C}_{14}$ ,  $\text{C}_{16}$  and  $\text{C}_{18}$  acids.

To our knowledge, this is the first case in which sphingosine derivatives have been found in plants. It is interesting to note the similarity of these compounds to caulerpicin (**3**), a mixture of  $N$ -acyl-2-aminohexadecan-1-ols from the green alga *Caulerpa racemosa* [5].



#### EXPERIMENTAL

PMR and  $\text{C}^{13}\text{NMR}$  spectra were determined on a 100 MHz spectrometer; chemical shifts are reported in  $\delta$  units (ppm) relative to TMS ( $\delta = 0$ ) as an int. stand. in  $\text{CDCl}_3$ . MS were obtained at 70 eV. Elemental analysis were performed by Chemical Analytical Services, University of California, Berkeley.

**Isolation of 2b.** *Laurencia nidifica* (575 g dry wt), collected near Port Allen, Kauai, was extracted successively with  $\text{Me}_2\text{CO}$  and hexane. The combined extracts were concd and partitioned between hexane and  $\text{H}_2\text{O}$ . The hexane soluble material (17.3 g) was chromatographed on a column of Si gel ( $62 \times 5$  cm). Eluting successively with solvents of gradually increasing polarity (hexane, THF,  $\text{Me}_2\text{CO}$ , MeOH) 12 fractions were collected. Fractions 9 and 10 (587 mg), eluted with  $\text{Me}_2\text{CO}$  and  $\text{Me}_2\text{CO}-\text{MeOH}$  (1:1) were combined and rechromatographed on neutral  $\text{Al}_2\text{O}_3$ . Elution with  $\text{CH}_2\text{Cl}_2-\text{MeOH}$  (97:3) removed 56 mg of a tan solid which was recrystallized from  $\text{CH}_3\text{CN}-\text{H}_2\text{O}$  to give 45 mg of (+)-2(S)- $N$ -acetamido-3(R)-acetoxyoctadecan-1-ol (**2b**) as white crystals, mp  $91.5-92.5^\circ$ ;  $[\alpha]_D + 10.8$  ( $\text{CH}_2\text{Cl}_2$ ,  $c$  1.25); PMR  $\delta$  0.85 (bt, C-Me), 1.23

(br s,  $-(\text{CH}_2)_n-$ ), 1.6 (m,  $-\text{CH}_2-\text{C}-\text{O}-$ ), 1.99 (s,  $-\text{NCOMe}$ ), 2.09 (s,  $-\text{OAc}$ ), 3.04 (br s,  $-\text{OH}$ ), 3.58 (bd,  $-\text{CH}_2\text{O}$ ), 4.02 (m,  $-\text{CH}-\text{N}$ ), 4.86 (bq,  $-\text{CH}-\text{O}$ ), 6.31 (bd,  $-\text{NH}-$ );  $\text{C}^{13}\text{NMR}$   $\delta$  172.17 (C=O), 170.32 (C=O), 73.97 (CH), 61.65 (CH), 53.46 ( $\text{CH}_2$ ), 31.97 ( $\text{CH}_2$ ), 31.35 ( $\text{CH}_2$ ), 29.77 (10  $\text{CH}_2$ ), 25.71 ( $\text{CH}_2$ ), 23.43 ( $\text{CH}_2$ ), 22.81 (Me), 21.05 (Me), 14.18 (Me); IR (KBr) 3380, 3280, 3060, 1729, 1642, 1540, 1235  $\text{cm}^{-1}$ ; MS  $m/e$  386 ( $\text{M} + 1$ ) 354, 324, 295, 294, 270, 157, 145, 102; high resolution MS  $m/e$  386.324979 (calcd for  $\text{C}_{22}\text{H}_{44}\text{NO}_4$ , 386.3271). Analysis. Found C, 68.35; H, 11.12; N, 3.56. Calcd. for  $\text{C}_{22}\text{H}_{43}\text{NO}_4$ : C, 68.53; H, 11.24; N, 3.63.

**Acetylation of 2b.** Acetylation of 19 mg of **2b** with  $\text{Ac}_2\text{O}$  and Py and recrystallization of the product from MeCN gave 11.2 mg of dihydrosphingosine triacetate (**2c**) as white crystals, mp  $100.5-101^\circ$ ,  $[\alpha]_D + 15.7$  ( $\text{CHCl}_3$ ,  $c$  0.86) [lit. [4] mp  $102-103^\circ$ ,  $[\alpha]_D + 18.1$  ( $\text{CHCl}_3$ ,  $c$  1.0)]; PMR  $\delta$  0.86 (bt, 3H), 1.23 (br s, 26 H), 1.6 (m, 2H), 1.98 (s, 3H), 2.04 (s, 3H), 2.06 (s, 3H), 3.96-4.52 (m, 3H), 4.87 (m, 1H), 5.88 (b, 1H).

**Isolation of 1b.** *Amansia glomerata* was collected at Black Point, Oahu and dried at  $40^\circ$ . The dried seaweed (1.35 kg) was extracted successively with petrol and  $\text{Me}_2\text{CO}$ . The  $\text{Me}_2\text{CO}$  extract (4.5 g) was chromatographed on neutral  $\text{Al}_2\text{O}_3$  with a solvent gradient of hexane,  $\text{CHCl}_3$  and MeOH. 9 fractions were collected. Fractions 4 and 5 eluted with  $\text{CHCl}_3-\text{MeOH}$  (99:1), were combined (115 mg). Filtration through Sephadex LH-20 ( $\text{CHCl}_3-\text{MeOH}$ , 3:2) and recrystallization from MeCN-EtOAc gave a mixture of  $N$ -acylsphingosines (**1b**) as an amorphous white solid, mp  $78-81^\circ$   $[\alpha]_D - 5.0^\circ$  ( $\text{CHCl}_3$ ,  $c$  1.2).

**Hydrolysis of 1b.** The mixture of  $N$ -acylsphingosines (**1b**, 63 mg) was hydrolyzed with 30 ml 1.2 M  $\text{H}_2\text{SO}_4$  in 85% MeOH at reflux for 4 hr [6]. The fatty acid Me esters (43 mg) were extracted from the hydrolyzate with heptane and the aq. phase was then basified with 2M NaOH and extracted with heptane- $\text{CH}_2\text{Cl}_2$  (1:1) to obtain 12 mg of crude sphingosine. The fatty acid esters were separated into two fractions by gel filtration on Sephadex LH-20. One fraction (34 mg) contained  $n\text{-C}_{23}$  (MS  $m/e$  354, 323),  $n\text{-C}_{24}$  ( $m/e$  368, 337), and  $n\text{-C}_{25}$  ( $m/e$  382, 351) acid Me esters in a ratio of ca 4:1:2; the other fraction (9 mg) contained tetradecanoic ( $m/e$  242, 211), ( $m/e$  270, 239) and octadecanoic ( $m/e$  298, 267) acid Me esters in a ratio of ca 5:6:1. The crude sphingosine was converted to the triacetate ( $\text{Ac}_2\text{O}$ , NaOAc). The product (11 mg)  $[\alpha]_D - 8.4$  ( $\text{CHCl}_3$ ,  $c$  0.83) [lit. [4]  $[\alpha]_D - 11.7$  ( $\text{CHCl}_3$ ,  $c$  1.0)] gave a PMR spectrum which was identical to that of authentic sphingosine triacetate.  $\delta$  0.84 (bt, 3H), 1.24 (br s, 22H), 1.4 (m, 2H), 1.96 (s, 3H), 2.04 (s, 3H), 2.05 (s, 3H), 3.9-4.6 (m, 3H), 5.1-5.95 (m, 4H).

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